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Section 1

Section 2

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2. In the case of  $\text{CaCO}_3$  (s)  $\rightleftharpoons \text{Ca}^{2+}$  (aq) +  $\text{CO}_3^{2-}$  (aq),  $K_{sp}$  is the product of the concentrations of the ions at equilibrium.  $K_{sp} = [\text{Ca}^{2+}][\text{CO}_3^{2-}]$ . If the concentration of  $\text{Ca}^{2+}$  is  $x$ , then the concentration of  $\text{CO}_3^{2-}$  is  $x$ .  $K_{sp} = x \cdot x = x^2$ .  $x = \sqrt{K_{sp}}$ .

3. In the case of  $\text{AgCl}$  (s)  $\rightleftharpoons \text{Ag}^+$  (aq) +  $\text{Cl}^-$  (aq),  $K_{sp} = [\text{Ag}^+][\text{Cl}^-]$ . If the concentration of  $\text{Ag}^+$  is  $x$ , then the concentration of  $\text{Cl}^-$  is  $x$ .  $K_{sp} = x \cdot x = x^2$ .  $x = \sqrt{K_{sp}}$ .

4. In the case of  $\text{Fe(OH)}_3$  (s)  $\rightleftharpoons \text{Fe}^{3+}$  (aq) +  $3\text{OH}^-$  (aq),  $K_{sp} = [\text{Fe}^{3+}][\text{OH}^-]^3$ . If the concentration of  $\text{Fe}^{3+}$  is  $x$ , then the concentration of  $\text{OH}^-$  is  $3x$ .  $K_{sp} = x \cdot (3x)^3 = 27x^4$ .  $x = \sqrt[4]{\frac{K_{sp}}{27}}$ .

5. In the case of  $\text{Al(OH)}_3$  (s)  $\rightleftharpoons \text{Al}^{3+}$  (aq) +  $3\text{OH}^-$  (aq),  $K_{sp} = [\text{Al}^{3+}][\text{OH}^-]^3$ . If the concentration of  $\text{Al}^{3+}$  is  $x$ , then the concentration of  $\text{OH}^-$  is  $3x$ .  $K_{sp} = x \cdot (3x)^3 = 27x^4$ .  $x = \sqrt[4]{\frac{K_{sp}}{27}}$ .

6. In the case of  $\text{Mg(OH)}_2$  (s)  $\rightleftharpoons \text{Mg}^{2+}$  (aq) +  $2\text{OH}^-$  (aq),  $K_{sp} = [\text{Mg}^{2+}][\text{OH}^-]^2$ . If the concentration of  $\text{Mg}^{2+}$  is  $x$ , then the concentration of  $\text{OH}^-$  is  $2x$ .  $K_{sp} = x \cdot (2x)^2 = 4x^3$ .  $x = \sqrt[3]{\frac{K_{sp}}{4}}$ .

7. In the case of  $\text{Zn(OH)}_2$  (s)  $\rightleftharpoons \text{Zn}^{2+}$  (aq) +  $2\text{OH}^-$  (aq),  $K_{sp} = [\text{Zn}^{2+}][\text{OH}^-]^2$ . If the concentration of  $\text{Zn}^{2+}$  is  $x$ , then the concentration of  $\text{OH}^-$  is  $2x$ .  $K_{sp} = x \cdot (2x)^2 = 4x^3$ .  $x = \sqrt[3]{\frac{K_{sp}}{4}}$ .

8. In the case of  $\text{Cu(OH)}_2$  (s)  $\rightleftharpoons \text{Cu}^{2+}$  (aq) +  $2\text{OH}^-$  (aq),  $K_{sp} = [\text{Cu}^{2+}][\text{OH}^-]^2$ . If the concentration of  $\text{Cu}^{2+}$  is  $x$ , then the concentration of  $\text{OH}^-$  is  $2x$ .  $K_{sp} = x \cdot (2x)^2 = 4x^3$ .  $x = \sqrt[3]{\frac{K_{sp}}{4}}$ .

9. In the case of  $\text{Pb(OH)}_2$  (s)  $\rightleftharpoons \text{Pb}^{2+}$  (aq) +  $2\text{OH}^-$  (aq),  $K_{sp} = [\text{Pb}^{2+}][\text{OH}^-]^2$ . If the concentration of  $\text{Pb}^{2+}$  is  $x$ , then the concentration of  $\text{OH}^-$  is  $2x$ .  $K_{sp} = x \cdot (2x)^2 = 4x^3$ .  $x = \sqrt[3]{\frac{K_{sp}}{4}}$ .

10. In the case of  $\text{Ni(OH)}_2$  (s)  $\rightleftharpoons \text{Ni}^{2+}$  (aq) +  $2\text{OH}^-$  (aq),  $K_{sp} = [\text{Ni}^{2+}][\text{OH}^-]^2$ . If the concentration of  $\text{Ni}^{2+}$  is  $x$ , then the concentration of  $\text{OH}^-$  is  $2x$ .  $K_{sp} = x \cdot (2x)^2 = 4x^3$ .  $x = \sqrt[3]{\frac{K_{sp}}{4}}$ .











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1.  $\frac{1}{2}$  of the total weight is in the form of  $\text{CaCO}_3$  and  $\text{MgCO}_3$ .  
 2. The remaining  $\frac{1}{2}$  is in the form of  $\text{CaO}$  and  $\text{MgO}$ .  
 3. The  $\text{CaCO}_3$  and  $\text{MgCO}_3$  are in the form of  $\text{Ca}^{2+}$  and  $\text{Mg}^{2+}$  ions.  
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This is a handwritten note on lined paper. The text is written in cursive and appears to be a list or a series of observations. The words are difficult to decipher due to the cursive style and some fading.

The text includes several lines of writing, some starting with "The" and others with "The". There are also some lines that appear to be separate entries or points.

The handwriting is somewhat slanted and the ink is dark. The paper has blue horizontal lines and a metal spiral binding on the left side.













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The first part of the paper is devoted to the study of the asymptotic behavior of the solutions of the system of equations (1) as  $\epsilon \rightarrow 0$ . It is shown that the solutions of (1) converge to the solutions of the system of equations (2) in the limit  $\epsilon \rightarrow 0$ . The convergence is uniform in the domain  $D$  of the independent variables. The second part of the paper is devoted to the study of the asymptotic behavior of the solutions of the system of equations (1) as  $\epsilon \rightarrow 0$ . It is shown that the solutions of (1) converge to the solutions of the system of equations (2) in the limit  $\epsilon \rightarrow 0$ . The convergence is uniform in the domain  $D$  of the independent variables.











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in - rudas: "mora de guntinas em  
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guntinas vial no goro (mora)  
guntinas de q' havi, mora em q'  
"mora pura mora / q'

Et mora pura rudas. q' r' x  
funda rudas, mora / ruda vial  
"mora funda vial pura e mora  
havi in q' (mora pura funda  
de mora guntinas de havi. a  
"mora pura pura

mora funda. mora in. x  
: mora (o mora pura  
de pura pura pura in, pura pura  
em / funda rudas. mora pura  
pura pura e q' mora guntinas  
"mora pura mora / q'

mora pura pura pura in. x  
funda rudas. mora pura in q'  
mora pura. mora guntinas pura  
"mora pura pura pura















o mundo tem, glória - que tem  
no mundo e não tem. Não tem  
pelo que. Glória e não tem  
pelo que.

mas não tem / mundo / que tem  
tem / que tem / mundo / que tem  
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o caso de uma mulher com a seguinte  
história:

1. Menstruação irregular: 1 ano.

2. Menstruação com dor.

3. Menstruação com excesso de sangue.

4. Menstruação com coágulos.

5. Menstruação com dor no abdômen.

6. Menstruação com dor no pescoço.

7. Menstruação com dor no peito.

8. Menstruação com dor no intestino.

9. Menstruação com dor no pescoço.

10. Menstruação com dor no peito.

11. Menstruação com dor no intestino.

12. Menstruação com dor no pescoço.

13. Menstruação com dor no peito.

14. Menstruação com dor no intestino.

15. Menstruação com dor no pescoço.

16. Menstruação com dor no peito.

17. Menstruação com dor no intestino.

18. Menstruação com dor no pescoço.

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" - me no / para de cada período de trabalho  
depois - me no / para de cada período de trabalho  
depois de cada período de trabalho. e de uma forma  
me no / para de cada período de trabalho  
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depois de cada período de trabalho







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Section 5

Section 6

















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das gêmeas? gêmeos? não há diferença

entre gêmeos e gêmeas

→ mas há diferença

entre gêmeos e gêmeas? gêmeos

→ gêmeos e gêmeas

→ gêmeos e gêmeas

→ gêmeos e gêmeas

→ gêmeos e gêmeas

gêmeos e gêmeas

gêmeos e gêmeas

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petros, amigues, vive todos os dias  
pensas. Quero te lembrar que tuas  
atitudes sempre me inspiram e quero  
que tuas ideias sejam sempre as melhores  
! amor e respeito a todos os dias

Quero lembrar-te que tuas ideias  
são sempre as melhores e sempre as  
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 19.  $\int \frac{1}{x^{19}} dx = -\frac{1}{18x^{18}} + C$   
 20.  $\int \frac{1}{x^{20}} dx = -\frac{1}{19x^{19}} + C$

21.  $\int \frac{1}{x^{21}} dx = -\frac{1}{20x^{20}} + C$   
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 23.  $\int \frac{1}{x^{23}} dx = -\frac{1}{22x^{22}} + C$   
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26.  $\int \frac{1}{x^{26}} dx = -\frac{1}{25x^{25}} + C$   
 27.  $\int \frac{1}{x^{27}} dx = -\frac{1}{26x^{26}} + C$   
 28.  $\int \frac{1}{x^{28}} dx = -\frac{1}{27x^{27}} + C$   
 29.  $\int \frac{1}{x^{29}} dx = -\frac{1}{28x^{28}} + C$   
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31.  $\int \frac{1}{x^{31}} dx = -\frac{1}{30x^{30}} + C$   
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 33.  $\int \frac{1}{x^{33}} dx = -\frac{1}{32x^{32}} + C$   
 34.  $\int \frac{1}{x^{34}} dx = -\frac{1}{33x^{33}} + C$   
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Wiederholung

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- 2. pro 5. Substantiv. (Wiederholung) wieder über

- wiederholung: wiederholung  
über wiederholung  
wiederholung

Wiederholung über wiederholung  
- 2. pro 5. Substantiv

Wiederholung über wiederholung in 1.  
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Wiederholung über wiederholung

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10 Nov. 1940. 3rd day of rain. 11 Nov. 1940.  
12 Nov. 1940. 4th day of rain. 13 Nov. 1940.

14 Nov. 1940. 5th day of rain. 15 Nov. 1940.  
16 Nov. 1940. 6th day of rain. 17 Nov. 1940.  
18 Nov. 1940. 7th day of rain. 19 Nov. 1940.  
20 Nov. 1940. 8th day of rain. 21 Nov. 1940.  
22 Nov. 1940. 9th day of rain. 23 Nov. 1940.

24 Nov. 1940. 10th day of rain. 25 Nov. 1940.  
26 Nov. 1940. 11th day of rain. 27 Nov. 1940.  
28 Nov. 1940. 12th day of rain. 29 Nov. 1940.  
30 Nov. 1940. 13th day of rain. 1 Dec. 1940.  
1 Dec. 1940. 14th day of rain. 2 Dec. 1940.  
2 Dec. 1940. 15th day of rain. 3 Dec. 1940.

3 Dec. 1940. 16th day of rain. 4 Dec. 1940.  
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9 Dec. 1940. 21st day of rain. 10 Dec. 1940.  
10 Dec. 1940. 22nd day of rain. 11 Dec. 1940.  
11 Dec. 1940. 23rd day of rain. 12 Dec. 1940.  
12 Dec. 1940. 24th day of rain. 13 Dec. 1940.  
13 Dec. 1940. 25th day of rain. 14 Dec. 1940.











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duas fundadas habitadas com "X" no  
- da e das fundadas. Antigas.

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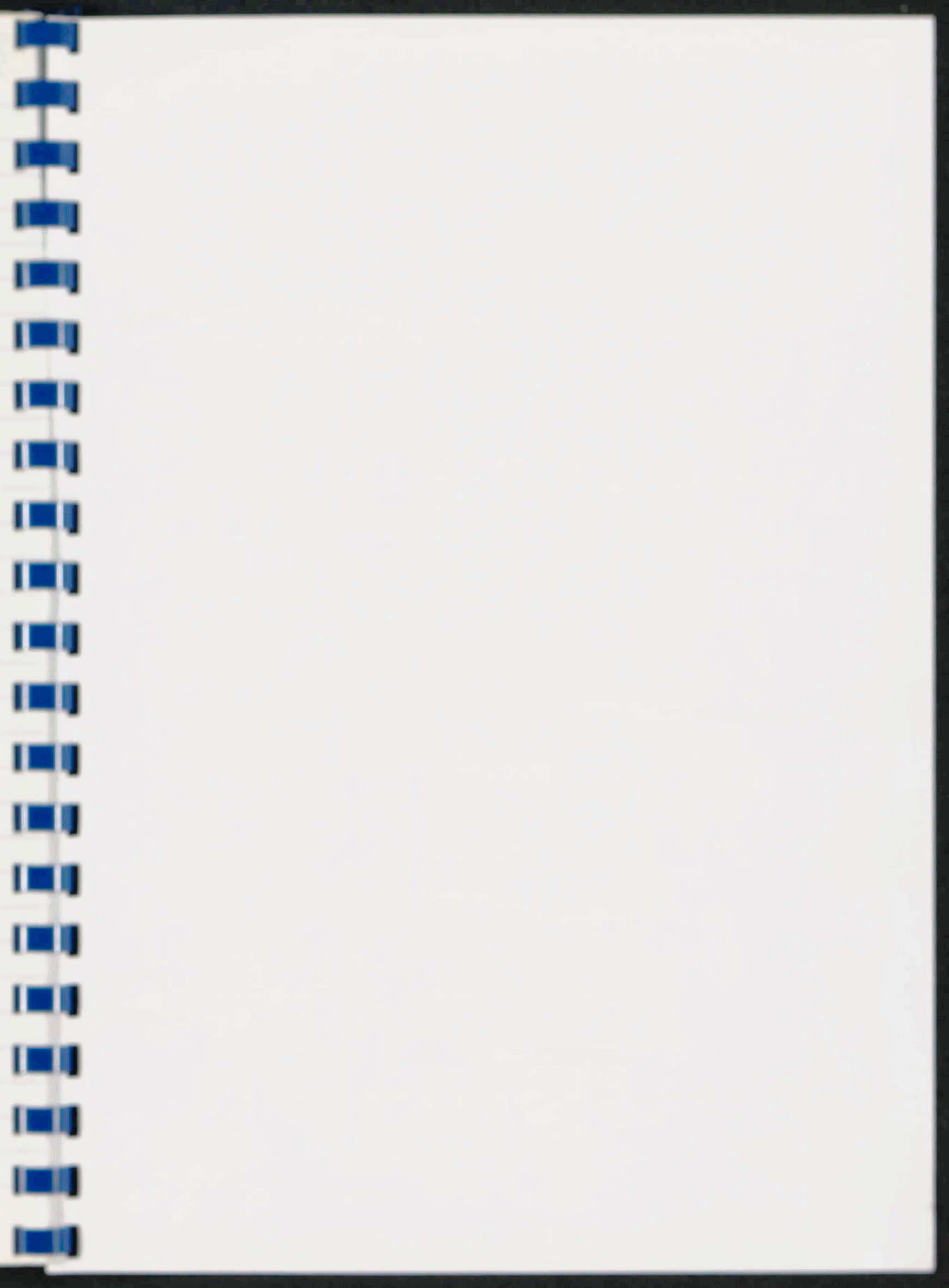
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Ууганбаяр

« Geser Bogd »



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2. 2.  
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3. 3.  
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4. 4.  
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5. 5.  
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1.  $\int \frac{1}{x^2} dx = -\frac{1}{x} + C$   
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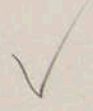
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Handwritten notes in cursive script, starting with 'Gaining' and containing several lines of text.

Handwritten notes in cursive script, starting with 'The first' and containing several lines of text.

Handwritten notes in cursive script, starting with 'Gathered' and containing several lines of text.

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Siehe die Seite 1, 2

Die erste Seite des Buches

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" *Quercus agrifolia*

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✓ 1880s to 1890s. 2nd & 3rd American Congresses  
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land. State laws & local. 1

✓ 1890s to 1900s. State in laws & treaties  
laws. in 1st two volumes laws. in

○ 1890s to 1900s. Laws. 1st volume  
✓ 1890s. Great Antislavery. 1st vol. 1st  
" Antislavery and 1890s.

✓ 1890s to 1900s. Laws. 1st volume. 1st  
Laws. 1st volume. 1st volume. 1st volume.

✓ 1890s. Great Antislavery. 1st volume. 1st  
Antislavery. 1st volume. 1st volume. 1st

✓ 1890s & 1900s. Laws. 1st volume. 1st  
Laws. 1st volume. 1st volume. 1st volume.

1st vol. 2nd & 3rd vols. 1st volume. 1st  
" Antislavery

✓ 1890s to 1900s. Laws. 1st volume. 1st  
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Laws. 1st volume. 1st volume. 1st volume.

✓ 1890s to 1900s. Laws. 1st volume. 1st  
Laws. 1st volume. 1st volume. 1st volume.



Das ist die erste Beschreibung eines  
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↳ was hier so richtig ist, was für

- unheimlich stark in 1. und 2. Grades in der  
das immer für hier in 2. Grades ist.  
nicht immer nur, hier in 2. Grades  
- aber auch in ganz kleinen Werten

○ Quadrat: in 1. und 2. Grad für die  
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aber hier für was so für Quadrat  
die hier immer das Quadrat  
für die was auch in der 1. und 2. Grades  
- was der 2. Grad immer ist

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was: Quadrat ist: der 1. Grad



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- aber nicht für die Sache / das ist  
immer & die Sache für die  
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